

Practice 3.2

Name _____

Period _____ Date _____

Symbol	Name	Atomic Number	Mass Number	Number of protons	Number of neutrons	Number of electrons
${}_{14}^{29}\text{Si}$						
	Cobalt-58					
		28	58			
		48			111	
				33	40	
					125	85
			226			88
${}_{79}^{197}\text{Au}$						
	Rhodium-103					
		80	201			
		78			117	
				82	125	
			118			50
					84	60
${}_{74}^{183}\text{W}$						
	Cesium-133					
		44	100			
		49			66	
				77	115	
			1			